

# A Fluorescence Method for Measuring the Retention of Coupling Factor (CF<sub>1</sub>) in Reconstitution Experiments of Photophosphorylation

Götz Harnischfeger and Reinhard Schopf

Lehrstuhl für Biochemie der Pflanze, Göttingen

(Z. Naturforsch. **32 c**, 392–397 [1977]; received January 26/February 16, 1977)

Fluorescence Label, Coupling Factor 1, Reconstitution, Photophosphorylation

The recombination of chloroplast coupling factor 1 (CF<sub>1</sub>) and thylakoid membranes in reconstitution experiments was studied through the fluorescence of paired labels covalently bound to NH<sub>2</sub>-groups. It was found that maximum recombination is achieved at a ratio of 1.5–3 µg CF<sub>1</sub>/µg chlorophyll. The addition of chloroplast lipids to the medium enhances the incorporation of CF<sub>1</sub> into the membranes.

The rates of ATP formation of the regenerated but labeled system are decreased to 50% of those found in unlabeled control experiments. This is discussed in context with the previous observation, that labeling of CF<sub>1</sub> at similar concentrations inhibits the ATPase activity of the isolated protein completely.

The possible use of double labeling in the study of the physical aspects of the reconstitution of the photophosphorylating system is discussed.

## Introduction

Important aspects of the mechanism of energy coupling in photophosphorylation can be studied by reconstitution experiments. In these, the coupling factor (CF<sub>1</sub>) is removed from the thylakoid membrane, thereby inactivating ATP formation, and subsequently added back to regenerate the system. The reconstitution attempts are generally successful, if thylakoid membranes are used which have not lost more than 30–50% of the native coupling factor by either EDTA treatment or pyrophosphate washing<sup>1–3</sup>. Such pretreated plastids lack, however, the capacity to maintain the high energy state and they show enhanced electron transport, two characteristics of uncoupling. It has been suggested that upon reconstitution of the photophosphorylating system, the renewed ability for ATP formation is solely provided by the previously present CF<sub>1</sub><sup>4</sup>.

The reconstitution experiment itself has two aspects. One is the purely physical problem of retention and integration of the added CF<sub>1</sub> into the thylakoid membrane. This structural aspect includes the parameters of interaction, *e. g.* binding sites, co-

factors of binding, the role of Mg<sup>2+</sup>, the specificity of CF<sub>1</sub> for combination with the thylakoid surface etc. The second concerns the mechanism by which the regenerated phosphorylating activity is maintained, *i. e.*, the functional aspect. An understanding of the latter is difficult without due regard to the former.

This paper concerns itself with a method which might give some information about the structural parameters involved in reconstitution experiments. It is based on a measurement of light energy transfer between fluorescent dye pairs, one attached to CF<sub>1</sub>, the other to the thylakoid membranes. Experiments in this direction have been reported earlier by Kraayenhof and Slater<sup>5</sup> using only fluorescamine to label covalently CF<sub>1</sub> and De Kouchkovsky<sup>6</sup> who coupled chloroplasts with fluorescein-isothiocyanate (FITC).

The concept of our experiments is simple. CF<sub>1</sub> and chloroplasts are coupled with different chromophores, whose optical properties are chosen in such a way, that the emission band of the first overlaps with the absorption band of the second. Transfer of excitation energy between the two chromophores indicates a close proximity of the labeled proteins. In this way it is possible to measure the retention parameters of CF<sub>1</sub> in reconstitution experiments even in the absence of functional capability for photophosphorylation. These dye pairs used in our experiments were fluorescamine/fluorescein and fluorescein/rhodamine.

**Abbreviations:** CF<sub>1</sub>, chloroplast coupling factor 1; DGD, digalactosyldiglyceride; FITC, fluorescein-isothiocyanate; MGD, monogalactosyldiglyceride; PC, phosphatidylcholine; PG, phosphatidylglycerol; PI, phosphatidylinositol; RITC, rhodamine-isothiocyanate; SL, sulfoquinovosyldiglyceride.

Requests for reprints should be sent to Priv. Doz. Dr. G. Harnischfeger, Untere Karspüle 2, D-3400 Göttingen.



Dieses Werk wurde im Jahr 2013 vom Verlag Zeitschrift für Naturforschung in Zusammenarbeit mit der Max-Planck-Gesellschaft zur Förderung der Wissenschaften e.V. digitalisiert und unter folgender Lizenz veröffentlicht: Creative Commons Namensnennung-Keine Bearbeitung 3.0 Deutschland Lizenz.

Zum 01.01.2015 ist eine Anpassung der Lizenzbedingungen (Entfall der Creative Commons Lizenzbedingung „Keine Bearbeitung“) beabsichtigt, um eine Nachnutzung auch im Rahmen zukünftiger wissenschaftlicher Nutzungsformen zu ermöglichen.

This work has been digitalized and published in 2013 by Verlag Zeitschrift für Naturforschung in cooperation with the Max Planck Society for the Advancement of Science under a Creative Commons Attribution-NoDerivs 3.0 Germany License.

On 01.01.2015 it is planned to change the License Conditions (the removal of the Creative Commons License condition "no derivative works"). This is to allow reuse in the area of future scientific usage.

This paper attempts an evaluation of the potential and shortcomings of this double labeling method.

### Material and Methods

CF<sub>1</sub> was isolated from spinach as described by Lien and Racker<sup>7</sup>. In some instances the method of Strotmann *et al.*<sup>8</sup> was used to obtain relatively pure preparations of CF<sub>1</sub>.

Chloroplasts, uncoupled but still retaining roughly 70–75% of their native CF<sub>1</sub>, were prepared according to Schopf *et al.*<sup>9</sup> by washing in 0.5 mM pyrophosphate or according to Strotmann *et al.*<sup>8</sup> by treatment with EDTA.

The reconstitution procedure of Strotmann *et al.*<sup>8</sup> was used with only slight modification. Surplus, non reacted CF<sub>1</sub> was removed by double washing of the plastids. The control experiments contained instead of CF<sub>1</sub> solution the same amount of buffer.

Emission spectra were measured using the non commercial fluorimeter described by Harnischfeger<sup>10</sup>. The assay contained in 3 ml suspension chloroplasts equivalent to 10–30 µg chlorophyll.

Photophosphorylation was measured in the PMS system either through incorporation of <sup>32</sup>P or by the phosphomolybdat complex in the usual way (Schopf<sup>11</sup>).

Red saturating light ( $\lambda > 600$  nm) was used in the measurements of the Hill reaction and of the proton gradient.

The dyes fluorescamine<sup>12</sup>, fluorescein- and rhodamine isothiocyanate<sup>13</sup> were used as covalent labels for both chloroplasts and CF<sub>1</sub>. Their absorption and emission bands overlap in the sequence fluorescamine (400–450 → 470–490), fluorescein (480–490 → 520–540), rhodamine (530–550 → 565–580). The shift in the maxima of absorption and fluorescence caused by covalent binding to proteins is relative small ( $\pm 15$  nm) and can be neglected in these experiments.

The covalent coupling of marker and protein presented no difficulties in the case of fluorescamine. The procedure of Udenfriend *et al.*<sup>12</sup> could be followed completely. On the other hand, for the reaction between the fluorescent isothiocyanates and protein Goldmann<sup>13</sup> recommends long incubation periods (6–12 h at 0 °C, and high pH 9.5). Since this diminishes or even abolishes activity, the procedure was modified to an incubation of 45 min at pH 8, compensating for the reduced reaction time by an increase of dye concentration to 50 µM. A sufficient though not complete labeling was thus accomplished. Unreacted marker was removed from chloroplasts by repeated washing, from CF<sub>1</sub> by gel-filtration through Sephadex G-25.

The electrophoretic behaviour of CF<sub>1</sub> was not influenced significantly by the covalent label, though its ability for activation by heat was markedly decreased, probably by side reactions between marker and DTT. For obvious reasons an activation by trypsin treatment was impossible to accomplish in tagged protein.

### Results and Discussion

The result of a reconstitution experiment in which EDTA treated thylakoids were labeled with FITC while CF<sub>1</sub> was marked with RITC is shown in Fig. 1. After recombination a marked increase in

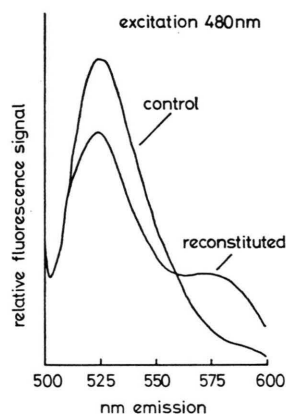


Fig. 1. Comparison of the emission spectrum of FITC labeled chloroplast membranes with that of the same preparation after reconstitution with RITC labeled CF<sub>1</sub>.

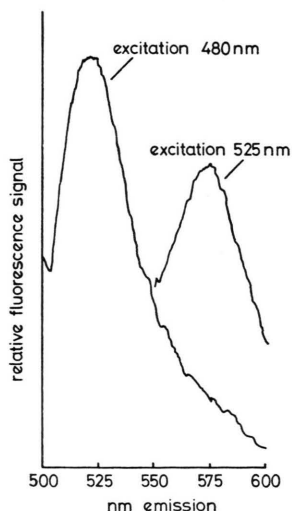


Fig. 2. Fluorescence emission spectra of a mixture of chloroplast membranes which were either labeled with FITC or RITC.

rhodamine fluorescence around 575 nm is observed when excited at 480 nm, while the emission of fluorescein, bound covalently to the chloroplasts, decreases.

In such experiments secondary effects have to be excluded. They can originate either in a slight absorption of RITC at 480 nm, or through the trivial effect, that energy emitted by fluorescein is reabsorbed by rhodamine. Such effects, however, can be neglected (Fig. 2). In this particular experiment a mixture of plastids was used, part of them labeled with RITC, the other half with FITC. The spectra can be resolved separately and do not show the secondary effects mentioned above.

The result in Fig. 1, thus, can be interpreted as a tight retention of CF<sub>1</sub> by the chloroplast membranes enabling energy transfer to occur. Such transfer is only possible if a minimum physical distance between the covalently bound pigments is achieved. It follows that the CF<sub>1</sub> has been incorporated into or attached itself tightly to the thylakoid membrane.

This explanation is strengthened by the fact, that several washes of the reconstituted system in reconstitution medium without CF<sub>1</sub> did not alter the energy transfer characteristics.

The validity of the conclusion is illustrated in Fig. 3, where a different combination of label was used. In this experiment CF<sub>1</sub> was labeled with fluorescamine while the thylakoid membrane carried covalently bound fluorescein. Excitation was at

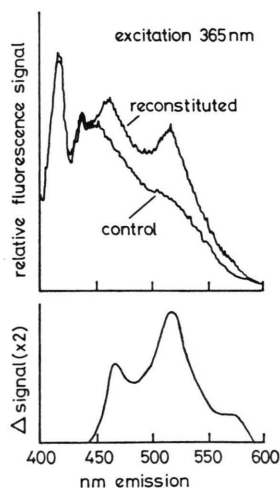


Fig. 3. Fluorescence emission spectra of FITC labeled thylakoid membranes before and after incubation with fluorescamine-labeled CF<sub>1</sub> in a reconstitution experiment. The resulting difference spectrum is shown in the lower half of the figure.

365 nm, the absorption region of fluorescamine. The retention of the protein after reconstitution is clearly in evidence. The difference spectrum shows an emission optimum of fluorescamine at 465 nm. The additional emission at 512–515 nm, due to fluorescein, shows the energy transfer between fluorescamine and fluorescein and indicates, thus, a tight attachment of CF<sub>1</sub> to the membrane. The two emission bands between 400 and 450 nm are instrumental artefacts due to the cuvette and the assay medium. They provide a convenient internal standard for normalization.

The interdependence of fluorescence emission and CF<sub>1</sub> retention is described in Fig. 4. As in all previous experiments surplus, non bound CF<sub>1</sub> was removed after reconstitution by several washings. If the fluorescence intensity is used as criterion for saturation, between 1.5–3  $\mu$ g CF<sub>1</sub> protein/ $\mu$ g chlorophyll suffices. This was confirmed in a series of experiments although there are differences between EDTA and pyrophosphate prepared CF<sub>1</sub> depleted thylakoids.

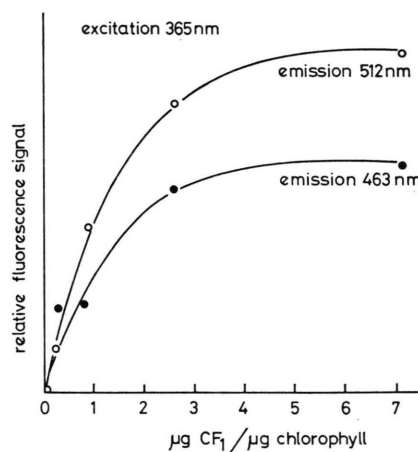


Fig. 4. Dependence of fluorescence emission of reconstituted thylakoids on the CF<sub>1</sub>/Chl ratio in the reconstitution assay. Excitation 365 nm, CF<sub>1</sub> was labeled with fluorescamine (F<sub>463</sub>), the thylakoids were labeled with FITC (F<sub>512</sub>).

A more general problem concerns the nature of the binding sites and the capacity of thylakoids to bind CF<sub>1</sub> in relation to phosphorylation activity. Strictly quantitative informations are unlikely since both EDTA and pyrophosphate treatment of chloroplasts releases other membrane components besides CF<sub>1</sub>. As shown previously by Heise and Jacobi<sup>14</sup> and Schopf *et al.*<sup>9</sup> a release (during pyrophosphate

and EDTA washings of plastids) and also a retention of lipids (during the reconstitution assay) takes place. To test their effect on the retention of CF<sub>1</sub>, reconstitution experiments were performed in the presence of isolated chloroplast lipids. The CF<sub>1</sub> used had been purified previously with diisopropylether and hexane in order to remove all lipids. Table I shows that under these circumstances an increase in fluorescence emission is observed. Since CF<sub>1</sub> was labeled with fluorescamine and the thylakoids with FITC the increase indicates a better incorporation of CF<sub>1</sub> into the membrane and a consequently higher energy transfer between both markers. It seems at a maximum if the sulfolipid fraction is used which is in accordance with the results of Heise<sup>15</sup> and Schopf *et al.*<sup>9</sup>.

Table I. Influence of lipids on the subsequently measured fluorescence emission of the labels in the recombined system. The lipids were added in an amount of 100  $\mu\text{g}$  each to the reconstitution assay of 0.55 ml. Excitation of the twice washed and resuspended (in 0.4 M sucrose — 5 mM MgCl<sub>2</sub> — 0.1 M Hepes pH 8) reconstituted thylakoids at 365 nm; 15  $\mu\text{g}$  Chl/measuring assay. The data have been normalized and the control arbitrarily set as 100. CF<sub>1</sub> labeled with fluorescamine (1.1 mg Prot/ml, 10 mM fluorescamine) thylakoids with FITC.

Addition	Relative signal		Ratio 517/463
	F <sub>463</sub>	F <sub>517</sub>	
—	100	100	1.0
DGD	258	238	.92
MGD	225	200	.89
PC+PI	276	223	.81
SL+PG	270	307	1.13

Direct information about the actual regeneration of phosphorylation activity, however, cannot be obtained through fluorescence measurements. The conclusions about spatial parameters obtained from the fluorescence experiments have always to be combined with functional data measured in the usual way.

To characterize the correlation between label and activity in more detail it has to be considered, that the label might influence the activity of the isolated components (CF<sub>1</sub> or thylakoids) in a different way than the recombined system.

No effect of the labels was found on the electron transport capacity of the chloroplast membranes used in the reconstitution experiments, which is in accordance with results reported by De Kouchkovsky<sup>6</sup> for FITC. It should be noted, however, that labeling of whole chloroplasts leads to a 50%

inhibition in the uncoupled state while basal electron transport is unaffected<sup>11</sup>.

The effect of the label fluorescamine on the ATPase activity of isolated CF<sub>1</sub> will be reported elsewhere<sup>16</sup>. Labeling of CF<sub>1</sub> by fluorescamine is complete at concentrations of 100–200  $\mu\text{M}$ . These amounts also resulted in a 90% inhibition of ATPase activity independent of the method of activation.

While the experiments of Schopf and Harnischfeger<sup>16</sup> argue for an almost complete deactivation of CF<sub>1</sub> by fluorescamine, the activity of the reconstituted CF<sub>1</sub>-thylakoid system is not affected this way.

Table II represents a comparison of the size of the proton gradient after reconstitution using unlabeled, partially CF<sub>1</sub> depleted thylakoids and CF<sub>1</sub>, marked prior to the experiment with increasing amounts of fluorescamine. A reduction of the size of the proton gradient is seen only at fluorescamine concentrations higher than necessary to achieve maximum fluorescence in the isolated CF<sub>1</sub>.

Table II. Influence of fluorescamine labeling of CF<sub>1</sub> on the functional regeneration of the proton gradient. The reconstitution assay contained in 1 ml 140  $\mu\text{g}$  chlorophyll and 260  $\mu\text{g}$  CF<sub>1</sub>, labeling previously with fluorescamine (235  $\mu\text{g}$  CF<sub>1</sub>/ml labeling assay).

$\mu\text{M}$ Fluorescamine in the labeling assay	Proton gradient ( $\mu\text{mol H}^+$ $\cdot \text{mg Chl}^{-1}$ )
0	299
50	318
100	321
250	276
500	184
Control no addition of CF <sub>1</sub>	96

The result indicates, that any regeneration of phosphorylating activity might rest in the 75% CF<sub>1</sub> not removed by the previous pyrophosphate or EDTA treatment of the chloroplasts used in the experiment. Assumed in this deduction is, however, an equivalence of the mechanism of ATP-ase activity *in vitro* and ATP formation *in vivo*.

Fig. 5 shows the dependence of the regenerated phosphorylation activity on the CF<sub>1</sub>/Chl ratio of the reconstitution assay. Unmarked CF<sub>1</sub> was recombined with FITC-labeled or unlabeled, CF<sub>1</sub>-depleted membranes. The labeled membranes show 30–50% less regeneration of phosphorylation activity than the unlabeled, control thylakoids. The rate of 150  $\mu\text{mol} \cdot \text{mg Chl}^{-1} \cdot \text{h}^{-1}$  is nevertheless considerable. The optimum CF<sub>1</sub>/Chl ratio is again around 1.5.



Fig. 6 shows the influence of the fluorescamine-concentration used to label CF<sub>1</sub> for the subsequent reconstitution experiment with either FITC coupled or unlabeled thylakoids. The CF<sub>1</sub>/Chl ratio in the assay was in the optimum range of 1.5. The data show, that the amount of fluorescamine label in CF<sub>1</sub> has no influence in the case of FITC-marked thyla-

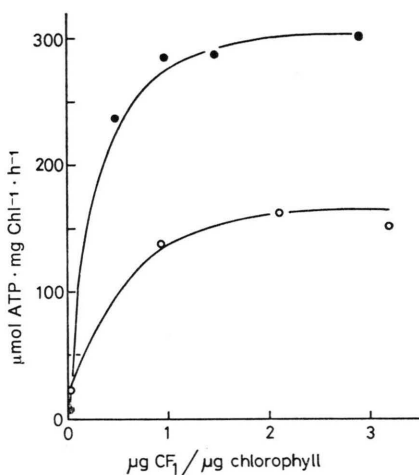


Fig. 5. Influence of the CF<sub>1</sub>/Chl ratio during reconstitution on the functional regeneration of cyclic photophosphorylation. Only the thylakoid membranes were labeled with FITC. The control (—●—) and also CF<sub>1</sub> were unlabeled.

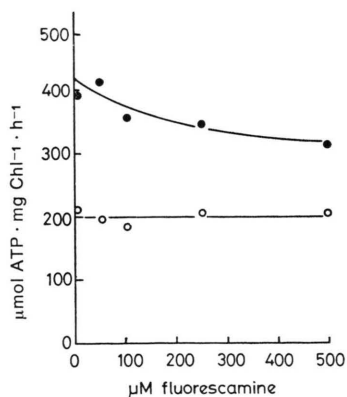


Fig. 6. Influence of CF<sub>1</sub> labeling with fluorescamine on subsequent regeneration of cyclic photophosphorylation. The thylakoid membranes of the reconstitution assay were either unlabeled (control —●—) or marked with FITC (—○—).

koids. A decrease around 30% is observed however in the control using unlabeled thylakoids.

The experiment argues that the amount of covalent label carried by and located exclusively at the freshly attached CF<sub>1</sub> must in some way influence the entire system.

This observation is not easy to interpret as a general inhibition of energy coupling between electron transport and CF<sub>1</sub>, especially since the labeled CF<sub>1</sub> is largely inactive<sup>16</sup> and the phosphorylation activity should rest in the residual factor carried by the thylakoid membrane. Experiments by Berzborn and Schröder<sup>23</sup> support the latter part of this argument.

An alternative explanation is the assumption of diminished mobility of CF<sub>1</sub> on or in the thylakoid membrane. Berzborn *et al.*<sup>17</sup> observed such a mobility which can be interpreted as a prerequisite for efficient phosphorylation. The observation that in Fig. 6 already a decrease in activity is observed if unlabeled CF<sub>1</sub> is attached to FITC labeled thylakoids, argues for this notion. It implies that the mobility and not alterations in CF<sub>1</sub> configuration<sup>5</sup> are the decisive factor under the experimental circumstances.

In summary, of the two parameters involved in an analysis of reconstitution the physical aspect seems to be well suited to fluorescence studies by the method of double labeling. Not only does the method allow direct measurements of a retention of CF<sub>1</sub> on or within the thylakoid membrane but it can provide also information about proximity of membrane and CF<sub>1</sub>-aminogroups. Further development might include an evaluation of conformational changes occurring in the membrane bound coupling factor which have been demonstrated through other means<sup>18–20</sup>. These can be determined conceivably through changes in fluorescence polarization *p* which is dependent on the conformation of the labeled species<sup>21,22</sup>. However, considerable theoretical work is necessary to develop the necessary equations applicable to such a complex system.

The technical assistance of Ms. S. Forbach is gratefully acknowledged. This study was supported by a grant from the Deutsche Forschungsgemeinschaft.

<sup>1</sup> V. Shoshan and N. Shavit, *Europ. J. Biochem.* **37**, 355–367 [1973].

<sup>2</sup> H. Hesse, R. Jank-Ladwig, and H. Strotmann, *Z. Naturforsch.* **31 c**, 445–451 [1976].

<sup>3</sup> N. Nelson, *Biochim. Biophys. Acta* **456**, 314–338 [1976].

<sup>4</sup> R. E. McCarty and E. Racker, *J. Biol. Chem.* **242**, 3435–3439 [1967].

<sup>5</sup> R. Kraayenhof and E. C. Slater, *Proc. IIIrd Int. Congr. Photosynthesis*, Rehovoth (M. Avron, ed.), p. 985–996, Elsevier, Amsterdam 1975.

- <sup>6</sup> Y. DeKouchkovsky, Proc. IIIrd Int. Congr. Photosynthesis, Rehovoth (M. Avron, ed.), p. 1013–1020, Elsevier, Amsterdam 1975.
- <sup>7</sup> S. Lien and E. Racker, Methods in Enzymology **XXIII** (A. San Pietro, ed.), p. 547–551, Academic Press, New York 1971.
- <sup>8</sup> H. Strotmann, H. Hesse, and K. Edelmann, Biochim. Biophys. Acta **314**, 202–210 [1973].
- <sup>9</sup> R. Schopf, K. P. Heise, B. Schmidt, and G. Jacobi, Proc. IIIrd Int. Congr. Photosynthesis, Rehovoth (M. Avron, ed.), p. 911–919, Elsevier, Amsterdam 1975.
- <sup>10</sup> G. Harnischfeger, Habilitationsschrift, Göttingen 1976.
- <sup>11</sup> R. Schopf, Dissertation, Göttingen 1976.
- <sup>12</sup> S. Udenfriend, S. Stein, P. Böhlen, W. Dairman, W. Leimgruber, and W. Weigele, Science **178**, 871–872 [1972].
- <sup>13</sup> M. Goldmann, Fluorescent Antibody Methods, Academic Press, New York 1968.
- <sup>14</sup> K.-P. Heise and G. Jacobi, Z. Naturforsch. **28 c**, 120–127 [1973].
- <sup>15</sup> K.-P. Heise, Abstr. Vorträge Dtsch. Bot. Ges., Würzburg 1974.
- <sup>16</sup> R. Schopf and G. Harnischfeger, Z. Naturforsch., in press.
- <sup>17</sup> R. J. Berzborn, F. Kopp, and K. Mühlethaler, Proc. IIIrd Int. Congr. Photosynthesis, Rehovoth (M. Avron, ed.), p. 809–820, Elsevier, Amsterdam 1975.
- <sup>18</sup> I. J. Ryrie and A. T. Jagendorf, J. Biol. Chem. **246**, 3771–3774 [1971].
- <sup>19</sup> D. A. Harris and E. C. Slater, Biochim. Biophys. Acta **387**, 335–348 [1975].
- <sup>20</sup> H. Strotmann, S. Bickel-Sandkötter, K. Boden, Kurzfassung, pg. 53 DBG, VAB-Botaniker Tagung, Zürich 1976.
- <sup>21</sup> G. Weber, Adv. Prot. Chem. **8**, 415–459 [1953].
- <sup>22</sup> W. B. Dandliker and V. A. DeSaussure, Immunochemistry **7**, 799–828 [1970].
- <sup>23</sup> R. Berzborn and P. Schröer, FEBS-Letters **70**, 271–275 [1976].